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| 10/521,176 | 01/14/2005 | Hiroshi Okamura | OKAMURA6 | 2935 |
| | 7590 06/09/200 D NEIMARK, P.L.L.C | EXAMINER | | |
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Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

| | Application No. | Applicant(s) | | |
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| | 10/521,176 | OKAMURA ET AL. | | |
| Office Action Summary | Examiner | Art Unit | | |
| | NARAYAN K. BHAT | 1634 | | |
| The MAILING DATE of this communication ap Period for Reply | pears on the cover sheet with the c | correspondence address | | |
| A SHORTENED STATUTORY PERIOD FOR REPL WHICHEVER IS LONGER, FROM THE MAILING D. - Extensions of time may be available under the provisions of 37 CFR 1. after SIX (6) MONTHS from the mailing date of this communication. - If NO period for reply is specified above, the maximum statutory period. - Failure to reply within the set or extended period for reply will, by statut Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b). | DATE OF THIS COMMUNICATION 136(a). In no event, however, may a reply be tin will apply and will expire SIX (6) MONTHS from e, cause the application to become ABANDONE | N. nely filed the mailing date of this communication. D (35 U.S.C. § 133). | | |
| Status | | | | |
| Responsive to communication(s) filed on 14 M This action is FINAL . 2b) ☑ This 3) ☐ Since this application is in condition for allowed closed in accordance with the practice under | s action is non-final. ance except for formal matters, pro | | | |
| Disposition of Claims | | | | |
| 4) ☐ Claim(s) 17-21,31 and 32 is/are pending in the 4a) Of the above claim(s) is/are withdra 5) ☐ Claim(s) is/are allowed. 6) ☐ Claim(s) 17-21, 31 and 32 is/are rejected. 7) ☐ Claim(s) is/are objected to. 8) ☐ Claim(s) are subject to restriction and/or | awn from consideration. | | | |
| Application Papers | | | | |
| 9) The specification is objected to by the Examina 10) The drawing(s) filed on is/are: a) accomposed and applicant may not request that any objection to the Replacement drawing sheet(s) including the correct to be a composed and the correct to be a correct | cepted or b) objected to by the lead of a drawing of the held in abeyance. Section is required if the drawing (s) is objection is | e 37 CFR 1.85(a). jected to. See 37 CFR 1.121(d). | | |
| Priority under 35 U.S.C. § 119 | | | | |
| 12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f). a) All b) Some * c) None of: 1. Certified copies of the priority documents have been received. 2. Certified copies of the priority documents have been received in Application No 3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)). * See the attached detailed Office action for a list of the certified copies not received. | | | | |
| Attachment(s) 1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review (PTO-948) 3) Information Disclosure Statement(s) (PTO/SB/08) Paper No(s)/Mail Date | 4) Interview Summary Paper No(s)/Mail Da 5) Notice of Informal F 6) Other: | ate | | |

Art Unit: 1634

Continued Examination Under 37 CFR 1.114

1. A request for continued examination under 37 CFR 1.114, including the fee set forth in 37 CFR 1.17(e), was filed in this application after final rejection. Since this application is eligible for continued examination under 37 CFR 1.114, and the fee set forth in 37 CFR 1.17(e) has been timely paid, the finality of the previous Office action has been withdrawn pursuant to 37 CFR 1.114. Applicant's submission filed on May 14, 2009 has been entered.

Status of the Claims

- 2. This action is in response to papers filed on May 14, 2009.
- 3. Claims 20, 31 and 32 were amended and claim 22 was cancelled. Claim amendments have been reviewed and entered.
- 4. Claims 17-21 and 31-32 are pending in this application and are under prosecution.
- 5. Applicant's arguments filed on May 14, 2009 have been fully considered and addressed following rejections.

Note about Priority

6. In view of the claimed priority to application Japan 2002-207886 and Japan2002-275797 filed in Japan and certified English translation of the application on May 14, 2009, the effective filing date for the instant claims is July 17, 2002.

Consolandi et al and Mao et al (WO 2003/020426) references are withdrawn in view of the effective filing date of the instant application.

Claim Rejections - 35 USC § 102

7. The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

- (a) the invention was known or used by others in this country, or patented or described in a printed publication in this or a foreign country, before the invention thereof by the applicant for a patent.
- (b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.
- (e) the invention was described in (1) an application for patent, published under section 122(b), by another filed in the United States before the invention by the applicant for patent or (2) a patent granted on an application for patent by another filed in the United States before the invention by the applicant for patent, except that an international application filed under the treaty defined in section 351(a) shall have the effects for purposes of this subsection of an application filed in the United States only if the international application designated the United States and was published under Article 21(2) of such treaty in the English language.
- 8. Claims 17-19 and 31 and 32 are rejected under 35 U.S.C. 102(b) as being anticipated by Jordan et al (Anal. Chem., 1997, 69, 4939-4947).

Claim 17 recites following structural components: a) a substrate, b) positively charged electrostatic layer on the substrate, c) a chemically modifying layer on the electrostatic layer, d) a nucleic acid molecule covalently bonded to the chemically modifying layer. Jordan et al teaches structural components 'a' to 'd' as described below.

Regarding structural component 'a', Jordan et al teaches a substrate (Fig. 2, substrate labeled as Au, pg. 4941, Experimental Section, subsection –Surface Attachment Chemistry, lines 1-3).

Regarding structural component 'b', Jordan et al teaches an electrostatic layer, i.e., poly-L-lysine layer comprising a positively charged amino group compound on the substrate (Fig. 2, electrostatic layer labeled as PL, pg. 4941, Experimental Section, subsection –Surface Attachment Chemistry, lines 8-11).

Regarding structural component 'c', Jordan et al teaches a chemically modifying layer, i.e., SSMCC layer on the electrostatic layer (Fig. 2, middle panel, pg. 4941, Experimental Section, subsection –Surface Attachment Chemistry, lines 11-16) making it possible to introduce a maleimide function group capable of covalently binding to a nucleic acid molecule (Fig. 2, middle panel).

Regarding structural component 'd', Jordan et al teaches a nucleic acid molecule bonded to the chemically modifying layer (Fig. 2, right panel, pg. 4941, Experimental Section, subsection –Surface Attachment Chemistry, lines 11-16).

Regarding claim 18, Jordan et al teaches that chemically modifying layer (i.e., SSMCC layer) contains a carboxyl group (Fig. 2, middle panel, pg. 4941, Experimental Section, subsection Materials, lines 2-3).

Regarding claim 19, Jordan et al teaches that electrostatic layer comprising poly L-lysine has non-covalent amino group distal to the substrate (Fig. 2, left panel).

Therefore, electrostatic layer include an amino group containing compound that does not covalently bound to the substrate.

Regarding claim 31, Jordan et al teaches that nucleic acid molecule is immobilized as a spot (Fig. 5).

Regarding claim 32, Jordan et al teaches that the thickness of the polylysine layer, i.e., an electrostatic layer is 27 Angstrom (i.e., 2.7 nm), which is in the range of 1 nm to 500 microns as claimed (Table 2).

9. Claims 17, 19-20 and 31 are rejected under 35 U.S.C. 102(a) and 102 (e) as being anticipated by Iwaki et al (USPN 6,858,392, effective filing date Dec. 13, 2000). *Note:* The 102(a) rejection is based on Iwaki et al (EP 1215287, published Jun. 19, 2002) with an earlier publication date of June 19, 2002. The subject matter of '287 publication is deemed to be identical because of claimed priority to JP 2000-379332 application filed in Japan on December 13, 2000. The citation used in the following rejection is based on the '392 patent.

Regarding structural component 'a', Iwaki et al teaches a substrate (Fig. 1, # 1, column 4, line 35).

Regarding structural component 'b', Iwaki et al teaches an electrostatic layer, i.e., poly-L-lysine layer comprising a positively charged amino group compound on the substrate (Fig. 1, electrostatic layer labeled as X, column 5, lines 40-44).

Regarding structural component 'c', Iwaki et al teaches a chemically modifying layer, i.e., E-G layer on the electrostatic layer (Fig. 1, panel two from the top, column 4, lines 35-41) making it possible to introduce a sulfonic function group capable of covalently binding to a nucleic acid molecule (column 4, lines 1-3 and 45-50).

Art Unit: 1634

Regarding structural component 'd', Iwaki et al teaches a nucleic acid molecule bonded to the chemically modifying layer (Fig. 1, panel 3 from the top, column 4, lines 1-3 and 45-50).

Regarding claim 19, Iwaki et al teaches that electrostatic layer comprising poly Llysine has non-covalent amino group distal to the substrate (Fig. 1, column 5, line 44).

Therefore, electrostatic layer include an amino group containing compound that does
not covalently bound to the substrate.

Regarding claim 20, Iwaki et al teaches a solid support wherein an electrostatic layer comprises an amino group-containing polymer, silane (Fig. 1, column 5, lines 50-53) and further teaches that the polymer binds covalently to the substrate and also has free amino group to introduce functional group (column 5, lines 50-56), which encompasses a compound containing an amino group at the terminus to which the substrate does not bind.

Regarding claim 31, Iwaki et al teaches that nucleic acid molecule is immobilized as a spot (Fig. 4, column 4, lines 1-3 and 45-50).

Claim Rejections - 35 USC § 103

10. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

- 11. This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).
- 12. Claims 17 and 20 are rejected under 35 U.S.C. 103(a) as being unpatentable over Jordan et al (Anal. Chem., 1997, 69, 4939-4947) in view of Iwaki et al (USPN 6,858,392, effective filing date Dec. 13, 2000).

Claim 20 is dependent from claim 17. Teachings of Jordan et al regarding claim 17 are described above in section 8.

Regarding claim 20, Jordan et al teaches a solid support wherein an electrostatic layer comprises an amino group-containing compound, i.e., polylysine (Fig. 2, left panel) and further teaches that the polylysine compound has an amino group at the terminus to which chemically modifying layer binds covalently (Fig. 2, middle panel), which encompasses a compound containing an amino group at the terminus to which the substrate does not bind. Jordan et al do no teach that the amino group containing compound covalently binding to the substrate. However, amino group containing compound covalently binding to the substrate was known in the art at the time of the claimed invention was made as taught by lwaki et al.

Iwaki et al teaches a solid support wherein an electrostatic layer comprises an amino group-containing polymer, silane (Fig. 1, column 5, lines 50-53) and further teaches that the polymer binds covalently to the substrate and also has free amino group to introduce functional group (column 5, lines 50-56), which encompasses a compound containing an amino group at the terminus to which the substrate does not bind. Iwaki et al also teaches covalent bonding of amino group containing compound to the substrate provide stable and reliable fixation of probes to the substrate (column 5, lines 55-57).

It would have been prima facie obvious to one of ordinary skill in the art at the time the invention was made to modify the amino group containing compound of Jordan et al with the silane amino group containing compound of Iwaki et al with a reasonable expectation of success with the expected benefit of having an amino group containing compound for covalent binding to the substrate providing stable and reliable fixation of probes to the substrate as taught by Iwaki et al (column 5, lines 55-57).

13. Claims 17, 19 and 21 are rejected under 35 U.S.C. 103(a) as being unpatentable over Jordan et al (Anal. Chem., 1997, 69, 4939-4947) in view of Woo et al (USPN 5,929,194 issued July 27, 1999).

Claim 21 is dependent from claim 19, which is dependent from claim 17.

Teachings of Jordan et al regarding claims 17 and 19 are described above in section 8.

Regarding claim 21, Jordan et al teaches an amino group containing compound polylysine (Fig. 2, left panel). Jordan et al do not teach about amino group containing

compound polyarylamine. However, amino group containing compound polyarylamine was known in the art at the time of the claimed invention was made as taught by Woo et al, who teaches polyarylamine compound for coating substrates and forming films on the substrate carrying positive charges (column 4, lines 8-10). Woo et al further teaches that the coating with polyarylamine makes the support solvent resistant and useful for fluorescent coating and as a protective coating for electronic devices (column 4, lines 10-11, column 14, lines 13-17).

It would have been prima facie obvious to one of ordinary skill in the art at the time the claimed invention was made to modify the substrate of Jordan et al with the polyarylamine coated substrate of Woo et al with a reasonable expectation of success with the expected benefit of generating fluorescent and protective coatings for electronic devices as taught by Woo et al (column 14, lines 13-17).

14. Claims 17 and 18 are rejected under 35 U.S.C. 103(a) as being unpatentable over Iwaki et al (USPN 6,858,392, effective filing date Dec. 13, 2000) in view of Mao et al (USPGPUB 2003/0124332, effective filing date Aug. 28, 2001).

Claim 18 is dependent from claim 17. Teachings of Iwaki et al regarding claim 17, are described above in section 9.

Regarding claim 18, Iwaki et al teaches solid support comprising a carboxyl group (column 5, line 1 and column 6, line 10). Iwaki et al do not teach chemically modifying layer containing a carboxyl group. However, chemically modifying layer

containing a carboxyl group was known in the art at the time of the claimed invention was made as taught by Mao et al.

Mao et al teaches a solid support comprising a substrate and an electrostatic layer comprising a positively charged amino group compound on the substrate (Fig. 1F, electrostatic layer labeled as first layer, paragraph 0046). Mao et al teaches a chemically modifying layer, i.e., a second layer on the electrostatic layer comprising polyacrylic acid containing carboxyl functional groups (paragraphs 0046, and 0049) capable of covalently binding to a nucleic acid molecule (paragraphs, 0045, 0049 and 0105).

Mao et al also teaches that the solid support comprising multilayer coated materials have high density of functional groups, limited leaching and strong and specific binding ability to a variety of agents and used for a variety of applications (paragraph 0017).

It would have been prima facie obvious to one of ordinary skill in the art at the time the invention was made to modify the functional groups on the chemically modifying layer of the substrate of lwaki et al with multilayer substrate comprising carboxyl functional group of Mao et al with a reasonable expectation of success with the expected benefit of having a chemically modifying layer with high density of carboxyl functional groups with limited leaching and strong and specific binding ability to a variety of agents for a variety of applications as taught by Mao et al (paragraph 0017).

15. Claims 17, 19 and 21 are rejected under 35 U.S.C. 103(a) as being unpatentable over lwaki et al (USPN 6,858,392, effective filing date Dec. 13, 2000) in view of Woo et al (USPN 5,929,194 issued July 27, 1999).

Page 11

Claim 21 is dependent from claim 19, which is dependent from claim 17.

Teachings of Iwaki et al regarding claims 17 and 19 are described above in section 9.

Regarding claim 21, Iwaki et al teaches a variety of amino group containing compounds including polylysine and silane (column 5, lines 44 and 50-51). Iwaki et al do not teach about amino group containing compound polyarylamine. However, amino group containing compound polyarylamine was known in the art at the time of the claimed invention was made as taught by Woo et al, who teaches polyarylamine compound for coating substrates and forming films on the substrate carrying positive charges (column 4, lines 8-10). Woo et al further teaches that the coating with polyarylamine makes the support solvent resistant and useful for fluorescent coating and as a protective coating for electronic devices (column 4, lines 10-11, column 14, lines 13-17).

It would have been prima facie obvious to one of ordinary skill in the art at the time the claimed invention was made to modify the substrate of Iwaki et al with the polyarylamine coated substrate of Woo et al with a reasonable expectation of success with the expected benefit of generating fluorescent coatings, protective coatings for electronic devices as taught by Woo et al (column 14, lines 13-17).

16. Claims 17 and 32 are rejected under 35 U.S.C. 103(a) as being unpatentable over Iwaki et al (USPN 6,858,392, effective filing date Dec. 13, 2000) in view of Bertrand et al (Macromol. Rapid Commun., 2000, 21, 319-348).

Claim 32 is dependent from claim 17. Teachings of Iwaki et al regarding claim 17 are described above section 9.

Regarding claim 32, Iwaki et al teaches an electrostatic layer (Fig. 1, Top panel). Iwaki et al do not teach about the thickness of the electrostatic layer is 1 nm to 500 micron. However, the thickness of the electrostatic layer was known in the art at the time of the claimed invention was made as taught by Bertrand et al.

Bertrand et al teaches a solid support comprising an electrostatic layer, wherein the thickness of the layer is from few angstroms to micrometer (Fig. 1 and pg. 319, column 2, lines 4-5), which is in the claimed range of 1 nm to 500 micron. Bertand et al further teaches that the electrostatic layer is very stable against mechanical stress or solvents (pg. 325, column 1, lines 1-3).

It would have been prima facie obvious to one of ordinary skill in the art at the time the claimed invention was made to modify the thickness of electrostatic layer of lwaki et al with the electrostatic layer of micrometer thickness of Bertrand et al with a reasonable expectation of success with the expected benefit of having electrostatic layer, which is very stable against mechanical stress or solvents as taught by Bertrand et al (pg. 325, column 1, lines 1-3).

17. Claims 17-20 and 31 are rejected under 35 U.S.C. 103(a) as being unpatentable over Mao et al (USPGPUB 2003/0124332, effective filing date Aug. 28, 2001) in view of Mirus et al (WO 01/02538 published Jan. 11, 2001).

Note: Reference of Mao et al used in this rejection with an effective filing date of August 28, 2001 is a prior art because effective filing date of the instant application is July 17, 2002.

Claim 17 recites following structural components: a) a substrate, b) positively charged electrostatic layer on the substrate, c) a chemically modifying layer on the electrostatic layer, d) a nucleic acid molecule covalently bonded to the chemically modifying layer. Mao et al teaches structural components 'a' to 'd' except for explicitly teaching covalent linking of nucleic acids.

Regarding structural component 'a', Mao et al teaches a substrate (Fig. 1F and paragraph 0046, line 9).

Regarding structural component 'b', Mao et al teaches an electrostatic layer, i.e., first layer comprising a positively charged amino group compound on the substrate (Fig. 1F, # first layer, paragraphs 0046 and 0069).

Regarding structural component 'c', Mao et al teaches a chemically modifying layer, i.e., a second layer on the electrostatic layer (Fig. 1F, # second layer, paragraph 0046). Mao et al further teaches that second layer comprises polyacrylic acid (paragraph 0069) containing carboxyl functional groups (paragraph 0105) capable of covalently binding to a nucleic acid molecule (paragraph 0045). While, Mao et al

suggests bound nucleic acid, the reference does not specifically teach a nucleic acid molecule bonded covalently to the chemically modifying layer.

Regarding claim 18, Mao et al teaches the second layer, i.e., chemically modifying layer contains a carboxyl group (paragraph 0105).

Regarding claim 19, Mao et al teaches a solid support wherein the first layer, i.e., an electrostatic layer comprises an amino group-containing polymer, polylysine (paragraph 0069) and further teaches that the polymer binds to the substrate by an electrostatic interactions (paragraph 0018), thus teaching an amino group containing compound that does not covalently bind to the substrate.

Regarding claim 20, Mao et al teaches that the first layer, i.e., an electrostatic layer comprises an amino group-containing compound (paragraph 0069) and further teaches that the polymer binds to the substrate through covalent bonds (paragraph 0018). Mao et al also teaches that the first layer comprising of amino group forms an amide bond with the carboxyl group of the polymer of the second layer (paragraph 0099) thus teaching a compound containing an amino group at the terminus to which the substrate does not bind.

Regarding claim 31, Mao et al clearly suggest nucleic acid molecules are immobilized on the solid support (paragraph 0045) but do not teach about immobilizing as a spot.

Regarding claim 17, Mao et al suggests bound nucleic acid, however, the reference does not specifically teach a nucleic acid molecule bonded covalently to the chemically modifying layer.

However, a nucleic acid molecule bonded covalently to the chemically modifying layer and immobilizing nucleic acid as a spot were known in the art at the time of the claimed invention was made as taught by Mirus et al.

Mirus et al teaches a solid support for nucleic acid immobilization comprising a substrate (pg. 4, line 30) and further teaches a chemically modifying layer, polyanion layer making it possible to introduce a carboxyl functional group (pg. 5, lines 3-9). Mirus et al further teaches that a nucleic acid molecule bonded covalently to the chemically modifying layer (pg. 3, lines 7-10).

Mirus et al also teaches that the nucleic acid molecule is immobilized as a spot (pg. 10, lines 3 and 29-30).

Mao et al suggests the covalent binding of nucleic acids to the functional groups on the surface of the chemically modified layer comprising poly acrylic acids (pg. 8, lines 23-27). Mirus et al teaches the covalent bonding of nucleic acids to the carboxyl groups on the chemically modified layer and immobilizing nucleic acids as a spot on the substrate, thus meeting the limitation of structural components recited in claims 17 and 31.

Mirus et al also teaches that covalent binding of nucleic acids increases the concentration of the nucleic acids irrespective of their size on the support, forming a

three dimensional lattice and increases the sensitivity of target detection (Tables 1-4, pg. 15, lines 9-13).

It would have been prima facie obvious to one of ordinary skill in the art at the time the invention was made to modify the nucleic acid immobilization to the substrate of Mao et al with covalent attachment of nucleic acid with a carboxyl functional group on the substrate of Mirus et al with a reasonable expectation of success with the expected benefit of covalent binding of nucleic acids increasing the concentration of the nucleic acids irrespective of their size on the support, forming a three dimensional lattice and increasing the sensitivity of target detection as taught by Mirus et al (Tables 1-4, pg. 15, lines 9-13).

18. Claims 17, 19 and 21 are rejected under 35 U.S.C. 103(a) as being unpatentable over Mao et al (USPGPUB 2003/0124332, effective filing date Aug. 28, 2001) in view of Mirus et al (WO 01/02538 published Jan. 11, 2001) as applied to claims 17 and 19 as above and further in view of Woo et al (USPN 5,929,194 issued July 27, 1999).

Teachings of Mao et al and Mirus et al regarding claims 17 and 19 are described above in section 17.

Regarding claim 21, Mao et al teaches a variety of amino group containing compounds including polylysine (paragraph 0019). Mao et al and Mirus et al do not teach about the amino group containing compound polyarylamine. However, amino group containing compound polyarylamine was known in the art before the claimed invention was made as taught by Woo et al, who teaches polyarylamine for coating

substrates and forming films on the substrate carrying positive charges (column 4, lines 8-10). Woo et al further teaches that coatings with polyarylamine makes the support solvent resistant and useful as a fluorescent coating, as a protective coating for electronic devices (column 4, lines 10-11, column 14, lines 13-17).

It would have been prima facie obvious to one of ordinary skill in the art at the time the invention was made to modify the substrate of Mao et and Mirus et al with the polyarylamine coated substrate of Woo et al with a reasonable expectation of success with the expected benefit of generating fluorescent and protective coatings for electronic devices as taught by Woo et al (column 14, lines 13-17).

19. Claims 17 and 32 are rejected under 35 U.S.C. 103(a) as being unpatentable over Mao et al (USPGPUB 2003/0124332, effective filing date Aug. 28, 2001) in view of Mirus et al (WO 01/02538 published Jan. 11, 2001) as applied to claim 17 as above and further in view of Bertrand et al (Macromol. Rapid Commun., 2000, 21, 319-348).

Teachings of Mao et al and Mirus et al regarding claim 17 are described above in section 17.

Regarding claim 32, Mao et al teaches that that electrostatic layer comprises of varying thickness (paragraph 0102). Mao et al and Mirus et al do not teach about the thickness of the electrostatic layer is 1 nm to 500 micron. However, the thickness of the electrostatic layer was known in the art at the time of the claimed invention was made as taught by Bertrand et al.

Bertrand et al teaches a solid support comprising a electrostatic layer, wherein the thickness of the of the layer is from few angstroms to micrometer (Fig. 1 and pg. 319, column 2, lines 4-5) and further teaches that the electrostatic layer is very stable against mechanical stress or solvents (pg. 325, column 1, lines 1-3).

It would have been prima facie obvious to one of ordinary skill in the art at the time the claimed invention was made to modify the substrate of Mao et and Mirus et al with the electrostatic layer of micrometer thickness of Bertrand et al with a reasonable expectation of success with the expected benefit of having electrostatic layer, which is very stable against mechanical stress or solvents (pg. 325, column 1, lines 1-3).

Response to remarks from the Applicants

Claim rejections under 35 U.S.C. § 102(a)

20. Applicant's arguments filed May 14, 2009 with respect to claims 17-20 and 31 as being anticipated by Consolandi et al have been fully considered but are moot in view of withdrawn rejections (Remarks, pg. 5, paragraphs 1-2).

Claim rejections under 35 U.S.C. § 103(a)

21. Applicant's arguments filed May 14, 2009 with respect to claims 17-20, 22 and 31 as being unpatentable over Mao et al and Mirus et al have been fully considered but are moot in view of withdrawn rejections (Remarks, pgs. 5 and 6). Applicant's arguments regarding the teachings of Mao et al are directed to Mao et al is not a reference because effective filing date of the instant application is prior to the publication date of

Art Unit: 1634

Mao et al. As described above in section 17, the new reference of Mao et al is a prior art because effective filing date of Mao et al used in this office action is prior to the filing date of the instant application.

Applicant's remaining arguments regarding claims 21 and 32 (Remarks, pgs. 6-7) are not persuasive because Applicant's arguments are directed to Mao et al is not a prior art. These arguments are not persuasive for the same reasons as described above. Furthermore, Applicants have not traversed the teachings, suggestions and motivation of secondary references.

Conclusion

22. No claims are allowed.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Narayan K. Bhat whose telephone number is (571)-272-5540. The examiner can normally be reached on 8.30 am to 5 pm.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, James Schultz can be reached on (571)-272-0763. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Art Unit: 1634

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

Narayan K. Bhat

Examiner, Art Unit 1634

/JD Schultz/

Supervisory Patent Examiner, Art Unit 1635